

# Fundamental XPS Data\* to Assist Peak-fitting Elements, Binary Oxides and Chemical Compounds

<p>1 <b>H 1s</b> H<sub>2</sub><sup>°</sup> LiH</p>													<p>2 <b>He 1s</b> He<sup>°</sup>/Be He<sup>°</sup>/C</p>												
<p>3 <b>Li 1s</b> LiOH Li<sub>2</sub>O</p>													<p>4 <b>Be 1s</b> Be<sup>°</sup> BeO<sup>°</sup></p>												
<p>11 <b>Na 1s</b> Na<sup>°</sup> Na<sub>2</sub>O</p>													<p>12 <b>Mg 2p</b> Mg<sup>°</sup> MgO<sup>°</sup></p>												
<p>19 <b>K 2p3</b> K<sup>°</sup> K<sub>2</sub>O</p>													<p>20 <b>Ca 2p3</b> Ca<sup>°</sup> CaCO<sub>3</sub></p>												
<p>27 <b>Co 2p3</b> Co<sup>°</sup> Co<sub>3</sub>O<sub>4</sub></p>													<p>28 <b>Ni 2p3</b> Ni<sup>°</sup> NiO<sup>°</sup></p>												
<p>35 <b>Br 3d5</b> KBr</p>													<p>36 <b>Kr 3d5</b> Kr<sup>°</sup>/Be Kr<sup>°</sup>/C</p>												
<p>37 <b>Rb 3d5</b> Rb<sup>°</sup> RbOAc</p>													<p>54 <b>Xe 3d5</b> Xe<sup>°</sup>/Be Xe<sup>°</sup>/C</p>												
<p>55 <b>Cs 3d5</b> Cs<sup>°</sup> Cs<sub>2</sub>O</p>													<p>86 <b>Rn 4f7</b> Radioactive</p>												

Atomic Number of Aluminum: 13  
 Abbreviation for Elemental Aluminum: Al<sup>°</sup>  
 Al 2p3 BE of Al<sup>°</sup> under Native Oxide: 72.8 (0.62)  
 Al 2p3 FWHM of Al<sup>°</sup> under Native Oxide: (1.39)  
 C 1s BE of Principal Peak 10 hours after Ion Etch, Pure Al<sup>°</sup>: 285.1  
 Al 2p3 BE (2<sup>nd</sup> decimal value included) of Ion Etched, Pure Al<sup>°</sup>: 72.82 (0.41)  
 Al 2p3 FWHM of Ion Etched, Pure Al<sup>°</sup>: 530.8 (1.60)

Principal XPS Signal for Aluminum: Al 2p3  
 Pure Oxide - most common form - thermodynamically stable  
 Al 2p3 BE of Oxide Species in Pure Oxide: 74.4 (1.39)  
 Al 2p3 FWHM of Oxide Species in Pure Oxide: (1.39)  
 C 1s BE of Hydrocarbon Peak - if non-conductive, then defined = 285.0 eV  
 O 1s BE of Oxygen Species in Pure Oxide: 530.8 (1.60)  
 O 1s FWHM of Oxygen Species in Pure Oxide: (1.60)

The FWHM and BE values in this table were all obtained by one scientist using two SSI XPS systems with a theoretical resolution of 0.1 eV. Each system was equipped with a monochromatic Aluminum X-ray source which have a theoretical energy resolution limit of about 0.16 eV. The BEs for the ion etched elements are reliable secondary energy reference values with a standard deviation of 0.055 as measured by NIST. BEs for the chemical compounds should be accurate to ±0.15 eV.

\*The "XI Library of Monochromatic XPS Spectra" is the source of BEs and FWHMs shown here. (US registered copyright, © TX 4-560-881)

Non-conductive materials were referenced to adventitious hydrocarbon with C 1s BE at 285.0 eV. Energy resolution settings for pure oxide data gave FWHM <0.75 eV for Ag 3d5 of ion etched Ag<sup>°</sup>. All non-conductors were analyzed with the Flood-Gun Mesh-Screen 0.5-1.0 mm above the specimen. C 1s BEs for "hydrocarbons" were collected from hydrocarbon peak that formed on ion etched elements. Carbon from the cryo-pumped vacuum (3x10<sup>-9</sup> torr) was analyzed >10 hours after ion cleaning. Energy resolution settings for ion etched elements gave FWHM <0.50 eV for Ag 3d5 of ion etched Ag<sup>°</sup>. Calibration was: Cu 2p3 at 932.67 ±0.05 eV, Cu 3s at 122.45 ±0.05 eV and Au 4f7 at 83.98 eV.

The BEs of non-conductive materials are referenced to the hydrocarbon C 1s BE at 285.0 eV to match current day methods of charge referencing, but this method of charge referencing is not absolute. The C 1s BEs of the adventitious hydrocarbon components on various naturally formed, thin native oxides (metal signal visible) were measured to occur between 285.5 and 286.5 eV. The native oxides of Ag, Al, Be, Bi, Cd, Co, Ga, Ge, Hf, In, Lu, Mg, Ni, Pb, Pd, Sc, Si, Ta, Y, Zn and Zr give C 1s BEs that are 0.5-1.5 eV above 285.0 eV. Surface dipole moments are suspected to be the cause of this shift. BEs of native oxides for these metals are more correct. BEs from oxides marked with a small red dot "°" require further study.

Size and pricing for these tables:

Laminated 8.5x10" letter or A4 size: \$55 (includes shipping)  
 Laminated 11x17" tabloid or A3 size: \$110 + shipping cost  
 Flexible thick plastic 2 x 3 ft with grommets for hanging: \$150 + shipping cost  
 Flexible thick plastic 3 x 5 ft with grommets for hanging: \$225 + shipping cost  
 Flexible thick plastic 4 x 6 ft with grommets for hanging: \$300 + shipping cost

Shipping of large tables depends on destination and speed: ranges from \$20 to \$100.  
 Choices: US Mail Express, or DHL, or FedEx Payment: Credit card or bank-to-bank

Changes and improvements include:

- Correction of several metal oxide metal signal BEs
- Correction of BEs of several O 1s peak BEs
- Coloration of chart to enhance readability and reveal related elements.
- Highlighting of FWHM in blue lettering because these are the key to improved peak-fits.
- Addition of alkali oxides replacing alkali halides to match presence of other oxides.
- Marking of ionic oxides (red dot) that are expected to have C 1s BEs higher than 285.5 eV.
- Addition of various BEs and FWHMs for Sr, La, Ce, Ne Os, U, Na, K, Rb, Cs and Th.
- Introduction of C 1s BEs from Carbon recovered after ion etching and 10+ hr exposure to UHV.
- Addition of radioactivity labels (bottom right corner of each box if radioactive)
- Calibration and Std Dev info shown in bottom image below signature

Copyrights © 2007 XPS International LLC, all rights reserved.  
 XPS International LLC, 754 Leona Lane, Mountain View, California, 94040, USA

B. Vincent Crist  
 Updated: March 1, 2007

58 <b>Ce 3d5</b> Ce <sup>°</sup> CeO <sub>2</sub>		59 <b>Pr 3d5</b> Pr <sup>°</sup> Pr <sub>2</sub> O <sub>5</sub>		60 <b>Nd 3d5</b> Nd <sup>°</sup> Nd <sub>2</sub> O <sub>3</sub>		61 <b>Pm 4d5</b> Pm <sup>°</sup> Pm <sub>2</sub> O <sub>3</sub>		62 <b>Sm 4d5</b> Sm <sup>°</sup> Sm <sub>2</sub> O <sub>3</sub>		63 <b>Eu 4d5</b> Eu <sup>°</sup> Eu <sub>2</sub> O <sub>3</sub>		64 <b>Gd 4d5</b> Gd <sup>°</sup> Gd <sub>2</sub> O <sub>3</sub>		65 <b>Tb 4d5</b> Tb <sup>°</sup> Tb <sub>4</sub> O <sub>7</sub>		66 <b>Dy 4d5</b> Dy <sup>°</sup> Dy <sub>2</sub> O <sub>3</sub>		67 <b>Ho 4d5</b> Ho <sup>°</sup> Ho <sub>2</sub> O <sub>3</sub>		68 <b>Er 4d5</b> Er <sup>°</sup> Er <sub>2</sub> O <sub>3</sub>		69 <b>Tm 4d5</b> Tm <sup>°</sup> Tm <sub>2</sub> O <sub>3</sub>		70 <b>Yb 4d5</b> Yb <sup>°</sup> Yb <sub>2</sub> O <sub>3</sub>		71 <b>Lu 4f7</b> Lu <sup>°</sup> Lu <sub>2</sub> O <sub>3</sub>	
883.8 (3.2)	882.1 (2.0)	929.4 (2.96)	928.8 (3.2)	980.8 (2.9)	983.2 (2.7)	Radioactive	Radioactive	1081.1	1083.8 (4.3)	128.2	136.2 (3.6)	140.2	(1187.5) (5.4)	145.9	(149.5) (?)	152.4	(153.5) (1.9)	159.8	160.9 (2.0)	167.7	168.5 (1.9)	175.3	176.3 (1.1)	182.4	185.0 (2.7)	7.1 (0.46)	8.0 (2.5)
285.0	285.0	285.0	285.0	285.6	285.0	Radioactive	Radioactive	285.0	285.0	284.3	285.0	281.4	285.0	285.0	285.0	285.0	285.0	285.0	285.0	285.0	285.0	285.3	285.0	285.6	285.0	284.6	285.0
529.6 (2.00)	529.6 (2.00)	528.5 (1.2)	528.5 (1.2)	530.8 (1.3)	530.8 (1.3)	Radioactive	Radioactive	128.7	529.9 (1.6)	128.18	530.5 (1.4)	140.31	529.3 (1.5)	145.95	529.7 (1.5)	152.34	529.4 (1.7)	159.58	529.3 (1.6)	167.25	529.7 (1.5)	175.37	529.6 (1.6)	182.39	529.2 (1.5)	7.10 (0.69)	529.5 (1.4)
90 <b>Th 4d5</b> Th <sup>°</sup> ThO <sub>2</sub>		91 <b>Pa 4d5</b> Radioactive		92 <b>U 4f7</b> U <sup>°</sup> UO <sub>3</sub>		93 <b>Np</b> Radioactive		94 <b>Pu</b> Radioactive		95 <b>Am</b> Radioactive		96 <b>Cm</b> Radioactive		97 <b>Bk</b> Radioactive		98 <b>Cf</b> Radioactive		99 <b>Es</b> Radioactive		100 <b>Fm</b> Radioactive		101 <b>Md</b> Radioactive		102 <b>No</b> Radioactive		103 <b>Lr</b> Radioactive	
333.2 (1.7)	334.5 (1.6)	Radioactive		377.3 (1.4)	381.7 (1.4)	Radioactive		Radioactive		Radioactive		Radioactive		Radioactive		Radioactive		Radioactive		Radioactive		Radioactive		Radioactive		Radioactive	

